

**CHEM 101A – Introductory University Chemistry I**

3 credits, 4 hours lecture, 3 hours lab

Atoms and molecules, states of matter, chemistry of the elements, relevance and uses of elements and compounds.

*Prerequisite: CHEM 30 or equivalent.*

**Instructor**

Dr. Sorin Nita

Office: S209F

Phone: (780) 715-3924

Email: [sorin.nita@keyano.ca](mailto:sorin.nita@keyano.ca)

**Office Hours**

Monday	2:00 PM – 4:00 PM
Thursday	3:00 PM – 4:00 PM
Friday	10:00 AM – 12:00 PM

**Hours of Instruction**

Lecture:	Monday	1:00 PM – 2:00 PM	Room 228
	Thursday	11:00 AM – 1:00 PM	Room 228
	Friday	2:00 PM – 3:00 PM	Room 228
Laboratory:	Tuesday	9:00 AM – 12:00 PM	Chem 101X - Lab 236
	Tuesday	2:00 PM – 5:00 PM	Chem 101Y - Lab 236
	Wednesday	9:00 AM – 12:00 PM	Chem 101Z - Lab 236

**Required Resources**

1. **General Chemistry: Principles & Modern Applications**; Petrucci, Herring, Madura, Bissonnette; Pearson Canada Inc., Toronto, Ontario, 2017, 11<sup>th</sup> edition, ISBN 978-0-13-293128-1.  
*The 10<sup>th</sup> edition of this textbook is also acceptable.*
2. **Chemistry 101/103 Laboratory Manual**; Keyano College, 2018/2019 edition.  
*The old editions of the lab manual are not acceptable.*
3. **Student Lab Notebook with Permanent Binding**; Hayden-McNeil, Plymouth, Michigan, ISBN 978-1-930882-00-3
4. A non-programmable scientific calculator (Sharp EL-531, used for exams, is recommended).
5. Extra-long lab coat.

### Course Outcomes

Upon successful completion of this course, the student shall be able to:

- Perform chemical experiments using laboratory equipment, and apply safety procedures to ensure a safe working environment for oneself and co-workers
- Summarize various atomic models and explain the modern approach towards atomic structure (quantum mechanical atom)
- Describe electronic configurations and correlate the chemical properties of elements with their electronic structure
- Analyze chemical compounds using various chemical bonding theories and predict their molecular structure (VSEPR), hybridization (Valence Bond Theory) and electronic diagrams (Molecular Orbital Theory)
- Analyze acid-base and redox reactions, balance chemical equations, and perform stoichiometry calculations

### Evaluation

Assignments	10%
Laboratory	25%
Midterm 1 Exam	6%
Midterm 2 Exam	14%
Final Exam	45%
Total	100%

*A grade of C- is required for progression or transfer.*

**Students are required to attend and complete all labs.** Unexcused absence from any lab period or failure to submit a lab report may result in a failing grade in the course. If a lab is missed for a valid reason, a makeup lab may be attended to complete the work.

### Grading System

Descriptor	Alpha Grade	4.0 Scale	Percent	Rubric for Letter Grades
Excellent	A+	4.0	> 92.9	Work shows in-depth and critical analysis, well developed ideas, creativity, excellent writing, clarity and proper format.
	A	4.0	85 – 92.9	
	A-	3.7	80 – 84.9	
Good	B+	3.3	77 – 79.9	Work is generally of high quality, well developed, well written, has clarity, and uses proper format.
	B	3.0	74 – 76.9	
	B-	2.7	70 – 73.9	
Satisfactory <b>Progression</b>	C+	2.3	67 – 69.9	Work has some developed ideas but needs more attention to clarity, style and formatting.
	C	2.0	64 – 66.9	
	C-	1.7	60 – 63.9	
Poor <b>Minimum Pass</b>	D+	1.3	55 – 59.9	Work is completed in a general way with minimal support, or is poorly written or did not use proper format.
	D	1.0	50 – 54.9	
Failure	F	0.0	< 50	Responses fail to demonstrate appropriate understanding or are fundamentally incomplete.

**Proposed Schedule of Topics**

	<b>textbook chapters</b>
<b>1. ATOMIC STRUCTURE</b>	
• Nature of light, atomic spectra, Bohr model of atom	8.1-8.4
• Nature of matter, quantum mechanical model of atom	8.5-8.6
• Shapes and energies of hydrogen orbitals, electron spin	8.7-8.11
• Periodic table, trends in atomic properties (sizes, IE, EA)	9.1-9.6
<b>2. CHEMICAL BONDING</b>	
• Lewis structures, polar bonds, electronegativity, formal charges	10.1-10.4
• Resonance, octet rule exceptions	10.5-10.6
• VSEPR and molecular structure	10.7-10.9
• Valence bond theory, hybridization, multiple bonding	11.1-11.4
• Molecular orbital theory	11.5-11.6
<b>3. STATES OF MATTER</b>	
• Ideal gases, mixtures of gases, partial pressures, Dalton's law	6.1-6.6
• Kinetic molecular theory, real gases	6.7-6.9
• Relation of gases, liquids, solids with intermolecular forces	12.1
• Properties of liquids and solids	12.2-12.3
• Phase diagrams	12.4
• Crystal structures	12.5-12.6
<b>4. SOLUTION STOICHIOMETRY</b>	
• Aqueous solutions, precipitation reactions	5.1-5.2
• Acid-base reactions, Arrhenius definition	5.3
• Redox reactions, Oxidizing and reducing agents	5.4-5.7

**Please Note:**

Date and time allotted to each topic is subject to change.

**Performance Requirements****Student Responsibilities**

It is your responsibility as a student to contact the Office of the Registrar to complete the forms for Withdrawal or Change of Registration, and any other forms. Please refer to the list of important dates as noted in the Academic Schedule in the Keyano College credit calendar.

More specific details are found in the Student Rights and Student Code of Conduct section of the Keyano College credit calendar. It is the responsibility of each student to be aware of the guidelines outlined in the Student Rights and Student Code of Conduct Policies.

**Laboratory Safety**

In the science laboratories, safety is important.

Students must complete the *WHMIS for Students* online training course on Moodle before entering the science laboratories.

Students must comply with the mandatory laboratory safety rules for this course as provided in the laboratory manual. Failure to do so will result in progressive discipline such as a verbal warning, refused entry into the laboratory, or suspension from the College.

**Student Attendance**

Class attendance is useful for two reasons. First, class attendance maximizes a students' learning experience. Second, attending class is a good way to keep informed of matters relating to the administration of the course (e.g., the timing of assignments and exams). Ultimately, you are responsible for your own learning and performance in this course.

It is the responsibility of each student to be prepared for all classes. Students who miss classes are responsible for the material covered in those classes and for ensuring that they are prepared for the next class, including the completion of any assignments and / or notes that may be due.

**Academic Misconduct**

Students are considered to be responsible adults and should adhere to principles of intellectual integrity. Intellectual dishonesty may take many forms, such as:

- Plagiarism or the submission of another person's work as one's own
- The use of unauthorized aids in assignments or examinations (cheating)
- Collusion or the unauthorized collaboration with others in preparing work
- The deliberate misrepresentation of qualifications
- The willful distortion of results or data
- Substitution in an examination by another person
- Handing in the same unchanged work as submitted for another assignment
- Breach of confidentiality.

The consequences for academic misconduct range from a verbal reprimand to expulsion from the College. More specific descriptions and details are found in the Student Rights and Student Code of Conduct section of the Keyano College credit calendar. It is the responsibility of each student to be aware of the guidelines outlined in the Student Rights and Student Code of Conduct Policies.

In order to ensure your understanding of the concept of plagiarism, you must successfully complete the online tutorial found on [ilearn.keyano.ca](http://ilearn.keyano.ca). Then print the certificate, sign it, and show it to each of your instructors. Your course work will not be graded until you show this signed certificate.

**Specialized Supports**

The Student Academic Support Services (SASS) department: Accessibility Services, Skill Centre and Wellness Services, work together to support student success at Keyano College.

**Accessibility Services (CC167)** supports student success through group and individualized instruction of learning, study and test taking strategies, and adaptive technologies. Students with documented disabilities, or who suspect a disability, can meet with the Learning Strategists to discuss accommodation of the learning barriers that they may be experiencing. Students who have accessed accommodations in the past are encouraged to visit our office at their earliest opportunity to discuss the availability of accommodations in their current courses. Individual appointments can be made by calling 780-791-8934

**Skill Centre (CC119)** provides a learning space where students can gather to share ideas, collaborate on projects and get new perspectives on learning from our tutorial staff. Students visiting the centre have access to one-to-one or group tutoring, facilitated study groups, and assistance in academic writing. The Skill Centre's Peer Tutor program provides paid employment opportunities for students who have demonstrated academic success and want to share what they have learned. Tutoring is available free to any students registered at Keyano College on a drop in basis, from 9:00 am to 5:00 pm Monday through Friday. Additional evening hours are subject to tutor availability and are posted in the Skill Centre.

**Wellness Services (CC260)** offers a caring, inclusive, and respectful environment where students can access free group and individual support to meet academic and life challenges. Mental Health Coordinators offer a safe and confidential environment to seek help with personal concerns. The Mindfulness Room in CC260 is available as a quiet space for students to relax during regular office hours. Wellness Service welcomes students to participate in any of the group sessions offered throughout the academic year addressing such topics as Mindfulness and Text Anxiety. Individual appointments can be made by calling 780-791-8934.

**Please watch your Keyano email for workshop announcements from our Student Academic Support Services team.**